## **Aquatic Plant Extracts**

		Crops	
1	Iden	tification of Petit	ioned Substance
2 3	<b>Chemical Names:</b> Not applicable	11 12	Acadian™ Organic Powder 0.5-0.3-14 (OMRI, 2005)
4 5	Other Names:		CAS Numbers:
6 7	Kelp extracts		84775-78-0 (Extracts of Ascophyllum nodosum)
8 9	Trade Names: Alg-A-Mic		<b>Other Codes:</b> None
10	Stress-X Powder		
13	Chara	starization of Dal	titioned Substance
14 15	Chara	cterization of Pet	moned Substance
16			
17	Composition of the Substance:		
18			
19			ally found in aquatic plants, as well as breadkdown
20 21			cess. Common sea plants identified in the literature as (Ascophyllum nodosum) and Sea Bamboo (Ecklonia maximu
22			nents are most commonly derived from kelp, specificall
23	from Ascophyllum spp. and other seawe		
24			
25	Specific chemical ingredients are not li	sted in product in	formation available for most commercial aquatic plant
26			eins, lipids, sugars, amino acids, and nutrients, vitamins,
27			nalysis of liquid aquatic plant extracts, provided by
28			ne Original TAP Database Form (1995), listed the
29	following substances as components of	f aquatic plant ext	racts:
30	A • A • 1 1 · • • 1	1 • •	.1 .1 1
31 32 33			cid, methionine, glycine, ethanolamine, lysine, e, valine, leucine, isoleucine, argenine, tyrosine,
34 35	• Vitamins: carotene, fucoxanth E, K	in, thiamine (B1),	riboflavin (B2), pantotenic acid, niacin, B12, C, D3,
36 37	• <b>Phytohormones:</b> _glycosyl cyto	kining auving and	d gibberellins
38	• Thytohormones. <u>g</u> rycosyrcyto	KIIIIIS, duxiiis dik	
39	• Other components: polysaccha	arides (i.e., compl	ex sugars), betaines <sup>1</sup>
40	<b>L</b> 1 7		0 //
41	Kelp contains a wide range of naturally	y occurring plant	nutrients and trace minerals essential to plant growth,
42			tain carbohydrates (e.g., alginic acid and mannitol) that
43			e to crop plants. These components of aquatic plant
44 45	extracts also contribute to building soil	structure (Acadia	an Seaplants Limited, Undated).
45 46	Crytalining a class of alart house	aucount in a secoli-	alast autoreta harra harra nor arta dita harra harra (* -*-1
46 47			plant extracts, have been reported to have beneficial fruits or seed heads, synchronization of flowering withi
48			101; Baker, 1996). When used as a plant dip, cytokinins
.0	a nera, and deray ed decay of mature p.	Land (Durawing 20	or, barer, 1990j. When abed as a plant ap, cytokiillis

 $<sup>^{1}</sup>$  Many plants use betaines to regulate and maintain osmotic pressure in their cells. This allows the plant to achieve tolerance to water shortages or heavy salt stresses (Nishimura et al., 2001).

- may also reduce transplant stress (Hall, 1997). Although other plant hormones (e.g., auxins, gibberellins) are
   present in aquatic plant extracts (Allen et al., 2001; Baker, 1996), they are not likely to be present in concentrations
- 50 present in aquatic plant extracts (Allen et al., 2001; Baker, 1996), they are not likely to be present in concentration 51 sufficient to have noticeable effect on plant growth (Baldwin, 2001). For this reason, some marketed products
- 52 may be enhanced with synthetic hormones to ensure performance of the product (Baldwin, 2001).
- 53

## 54 **Properties of the Substance**:

55

56 Aquatic plant extracts are described as a brownish-black powder or brown liquid with a mild marine (or

- 57 fish) odor (Original TAP Database Form, 1995; Necessary Organics, 1991). The material is soluble in water
- and ethanol, is stable, and is not subject to hazardous polymerization (Original TAP Database Form, 1995;
- 59 Necessary Organics, 1991). It is non-flammable (Necessary Organics, 1991).
- 60

## 61 Specific Uses of the Substance:

62

According to the Original TAP Database Form (1995), aquatic plant extracts are used as foliar fertilizers (on all crops) or as soil conditioners. Aquatic plant extracts also are used in combinations as a foliar/soil feed

- 65 or transplant solution and seed treatment. The material is then absorbed into the plant and acts as a
- 66 growth promoter. Aquatic plant extracts can be used as a source of micronutrients and growth promoters;
- as a rooting solution for transplants and cuttings; for cold hardiness in tomato plant, citrus fruits, and
- cabbage; and to reduce pest damage (nematodes in tomato and okra; mites in strawberries, peaches and
- 69 apples) (Original TAP Database Form, 1995).
- 70

Aquatic plant extracts are sometimes applied as a foliar spray by farmers seeking a natural source of

- 72 micronutrients. For the most part, none of the micronutrient levels in kelp extracts are high enough to 73 correct a deficiency, but they are used as a "tonic" providing a broad array of micronutrients and other 74 trace elements in organic farming (Hall and Sullivan, 2001). Some aquatic plant extract products are 75 supplemented with synthetic nutrients (Hall and Sullivan, 2001).
- 76

## 77 Approved Legal Uses of the Substance:

- 78
- 79 <u>National Organics Program (NOP)</u>80

81 Aquatic plant extracts (other than hydrolyzed) are currently on the National List for use as a plant or soil

amendment. Annotations to the National List limit the allowable extraction process to those using

83 potassium hydroxide or sodium hydroxide solvents, and the solvent amount used is limited to that amount

- 84 necessary for extraction. Extraction processes are described further in Evaluation Question #1.
- 85 86 <u>EPA</u>
- 87

Extracts of *Ascophyllum nodosum* (CAS number 84775-78-0) are on EPA's list 4B of inert ingredients in
 pesticides.

90

EPA exempted cytokinin in aqueous extract of seaweed meal (as opposed to seaweed extract) from the requirement for a tolerance in all food commodities when used as a plant growth regulator on plants,

- seeds, or cuttings and on all food commodities after harvest (EPA, 1998).
- 94

95 <u>Action of the Substance</u>:96

- 97 As previously mentioned, aquatic plant extracts are used as foliar fertilizers or as soil conditioners. An
- 98 understanding of the mechanisms of action that produce the beneficial effects of aquatic plant extracts is
- 99 incomplete (Henry, 2005). Research has demonstrated that many crops do not respond to seaweed extract
- 100 when soil fertility and production conditions are favorable (Kuisma, 1989; McGeary and Birkenhead, 1984).
- 101 Other studies have shown that crops are more likely to respond to kelp products under stressful growing
- 102 conditions (Verkleij, 1992; Nelson and Van Staden, 1984; Abetz, 1980). However, a study in 1995 found no

103 104 105	beneficial effects to tomatoes of kelp-derived foliar sprays when the plants were inadvertently subjected to stress early in growth (Tourte, 1997).
106 107 108 109 110 111 112 113 114	Foliar fertilization is subject to variables associated with (1) uncertainty and variability of nutrient content of material sprayed and (2) leaf uptake processes (Tourte, 1997). More specifically, the product efficacy may be influenced by the particular species used to make the extract, the growth stage of the plants, the time of harvest and post-harvest handling of the plants used in the product, the processing method, and the product's shipment and storage conditions (Henry 2005). Effectiveness of kelp extracts probably varies with crop type, quality of the foliar spray itself, soil nutrient availability, timing of spray applications, and soil moisture and weather conditions. For some crop and/or conditions, foliar treatment with kelp products may have no effect (Tourte, 1997).
<ol> <li>115</li> <li>116</li> <li>117</li> <li>118</li> <li>119</li> <li>120</li> </ol>	Aquatic plant extracts used as a soil amendment improve soil fertility by making nutrients and hormones more directly available to plants. Aquatic plant extracts contain carbohydrates such as alginic acid and mannitol, which enable plants to better absorb nutrients from the soil. The carbohydrates are broken down resulting in the stimulation of beneficial soil bacteria that fix nitrogen making it available to plant roots. Furthermore, these activities decrease the need for chemical fertilizers (Smittle, 1991).
120 121 122 123 124 125 126	Additionally, hormones (e.g., auxins, cytokins, and gibberellins) contained in plant extracts are absorbed into the plant and act as growth promoters. When used as a plant growth regulator, seaweed extract would be expected to alter cell division, root and shoot elongation, initiation of flowering, and other metabolic functions (in contrast to fertilizers, which simply supply minerals needed for nutrition) (Allen et al., 2001).
127	Status
128	
129 130 131 132	<u>International</u> Canada - Canadian General Standards Board (CGSB) - http://www.pwgsc.gc.ca/cgsb/032_310/32.310epat.pdf
133 134 135 136 137	As of June 2004 draft (CGSB, 2004), aquatic plant extracts were accepted for use in Canada if they were natural (nonsynthetic). The only synthetic solvents allowed for extraction are potassium hydroxide or sodium hydroxide, which are not to exceed the amount necessary for extraction. If the product contains other synthetic preservatives (e.g., formaldehyde) or is fortified with otherwise prohibited plant nutrients, then the aquatic plant product is prohibited in Canada.
138 139 140	CODEX Alimentarius Commission - ftp://ftp.fao.org/docrep/fao/005/Y2772e/Y2772e.pdf
141 142 143	Based on Table 1 of the CODEX, substances for use in soil fertilizing and conditioning - seaweeds and seaweed products - need recognition by the certification body or authority.
144 145 146	European Economic Community (EEC) Council Regulation 2092/91 – http://europa.eu.int/eur-lex/en/consleg/pdf/1991/en_1991R2092_do_001.pdf
147 148 149	Aquatic plant extracts are allowed in the EU, following Annex IIB – Seaweed and seaweed products (Organic Trade Association, 2002).
150 151 152	Seaweeds and seaweed products are allowed for use as fertilizers and soil conditions, as far as directly obtained by:
152 153 154 155	<ul> <li>Physical processes including dehydration, freezing and grinding;</li> <li>Extraction with water or aqueous acid and/or alkaline solution; and</li> <li>Fermentation;</li> </ul>

157 Need recognition by the inspection body or inspection authority.

158	
159	Japan Agricultural Standard for Organic Production –
160	http://www.ams.usda.gov/nop/NOP/TradeIssues/JAS.html
161	
162	No information pertaining to aquatic plant extracts was located at this site.
163	
164	California Certified Organic Growers International (CCOF) -
165	http://www.ccof.org/pdf/GlobalMarketAccessDraftForReview.pdf
166	
167	No information pertaining to aquatic plant extracts was located at this site.
168	
169	Washington State Department of Agriculture: European Organic Verification Program (EOVP) -
170	http://agr.wa.gov/FoodAnimal/Organic/default.htm
171	
172	No information pertaining to aquatic plant extracts was located at this site.
173	
174	Evaluation Questions for Substances to be used in Organic Crop or Livestock Production
4	L'alantion Questions for Substances to be used in organice crop of Livestock ribuation
175	Francisco Occasión #1. In the metition of extension former lateral encourse for the extension lateral encourse?
176	<i>Evaluation Question #1:</i> Is the petitioned substance formulated or manufactured by a chemical process?
177	(From 7 U.S.C. § 6502 (21))
178	Alleali Extraction of Aquatic Plant Extractor
179 180	Alkali Extraction of Aquatic Plant Extracts:
	A scarding to the Original TAD Detahase Form (100E), secured outract is preduced from fresh live plants
181	According to the Original TAP Database Form (1995), seaweed extract is produced from fresh, live plants
182	which are processed into a soluble powder or liquid, and may be stabilized with synthetic acids and fortified with other incredients.
183	fortified with other ingredients.
184	A quatic plant extract manufacturers utilize an alkali extraction process to "direct" the plants and derive
185	Aquatic plant extract manufacturers utilize an alkali extraction process to "digest" the plants and derive
186	both micronutrients and naturally occurring plant hormones from plant sources. This process also
187	transforms the plants into a soluble, easily transported form. The majority of manufacturers use potassium
188	hydroxide as the primary reagent in the alkali extraction process. Other alkali reagents used by some
189	manufactures include sodium hydroxide, calcium hydroxide, and sodium carbonate (Henry, 2005).
190	All all states the second state is the set of the set of the set of the second states and second states and second states in the second states and second st
191	Alkali extraction may be conducted either at ambient or elevated temperatures and pressures. Some
192	manufacturers claim that there is a reduced yield of desired plant growth hormones when the products are
193	extracted at higher temperatures. However, no studies have been found comparing the effectiveness of
194	extracts produced at different concentrations of alkali, or at different temperatures (Henry, 2005).
195	Following all all dispetion must many fasterions and use the aTT of the analyst with an axid must commonly
196	Following alkali digestion, most manufacturers reduce the pH of the product with an acid, most commonly
197	phosphoric acid (Henry, 2005). Phosphoric acid is a synthetic material (Original TAP Review, 1999). Only
198	the amount of acid needed to achieve a pH of 3.5 is recommended (Conrad, 2002). It is not clear if
199	neutralization is a necessary step. One form of the most widely-used extract product (Maxicrop) is not
200	neutralized, and a 25 percent solution has a pH of 9-10. No studies have been found comparing the
201	effectiveness of high-pH versus neutralized alkali extracts (Henry, 2005).
202	Noncomplatic Draduction of Aquatic Diant Extracts
203	Nonsynthetic Production of Aquatic Plant Extracts
204	According to the Original TAP Database Form (100E), equation last outre to see also be deviced active 11-
205	According to the Original TAP Database Form (1995), aquatic plant extracts can also be derived naturally.
206	For example, kelp can be dehydrated after harvest by sun drying, and then ground into a meal product.
207	This kelp meal can be sprinkled directly on the soil, or diluted with water and either sprayed on plant
208	foliage as a foliar spray or poured directly into the ground as a soil drench (3R Lighting, 2005).
209	Nonsynthetic products also may be produced using mechanical disruption, or freezing, pulverization, and clarification of the themas durry (Honry 2005). The relative officeacy of alkali extracted versus per alkali
210	clarification of the thawed slurry (Henry, 2005). The relative efficacy of alkali-extracted versus non-alkali- extracted product has not been consistently domenstrated, perhaps partly as a result of a lack of
211	extracted product has not been consistently demonstrated, perhaps partly as a result of a lack of

212 213	understanding of the mechanism by which aquatic plant extracts exert any purported beneficial effect (Henry, 2005).
214 215 216 217 218	<u>Evaluation Question #2:</u> Is the petitioned substance formulated or manufactured by a process that chemically changes the substance extracted from naturally occurring plant, animal, or mineral sources? (From 7 U.S.C. § 6502 (21).)
219 220 221 222 223	Yes, some chemical reactions do occur during the processes by which aquatic plant extracts are obtained from plant sources, such as kelp and seaweed. However, because aquatic plant extracts are complex chemical mixtures, it may not be possible to characterize all the reactions and the extent to which there are chemical changes beyond the simple effects of shifting pH.
224 225 226 227 228 229	Alkali "digestion" of aquatic plants involves the breakdown of galactans in seaweed cell walls (Sideman, 2004). Galactans are tough, fibrous polymeric saccharides (primarily repeating units of galactose). With alkali extraction, this breakdown is accomplished through partial hydrolysis of the galactans, catalyzed by bases (such as sodium hydroxide, potassium hydroxide, potassium carbonate) used in sufficient quantity to achieve a pH of 9 - 10 (Sideman, 2004). Without alkali extraction this breakdown can be accomplished through non-chemical means such as freeze and thaw or grinding.
230 231 232 233	<u>Evaluation Question #3:</u> Is the petitioned substance created by naturally occurring biological processes? (From 7 U.S.C. § 6502 (21).)
233 234 235 236 237 238	Aquatic plant extracts contain substances that naturally occur in aquatic plants or that are produced when aquatic plants are broken down. As described in Evaluation Question #1, aquatic plant extracts can be produced using nonsynthetic methods. However, aquatic plant extracts produced by alkali digestion, as currently permitted by the National List, are not derived with a naturally occurring biological process.
239 240	<u>Evaluation Question #4:</u> Is there environmental contamination during the petitioned substance's manufacture, use, misuse, or disposal? (From 7 U.S.C. § 6518 (m) (3).)
<ul> <li>241</li> <li>242</li> <li>243</li> <li>244</li> <li>245</li> <li>246</li> </ul>	According to Tooby (2003) plant extracts appear to be biodegradable and are likely to have a low impact on crops or stored agricultural products. Additionally, there is no information available from EPA to suggest that environmental contamination results from the manufacture, use, misuse, or disposal of aquatic plant extracts. Environmental contamination could result from improper disposal of acids or bases used in alkali extraction.
247 248 249 250	Evaluation Question #5: Is the petitioned substance harmful to the environment? (From 7 U.S.C. § 6517 (c) (1) (A) (i) and 7 U.S.C. § 6517 (c) (2) (A) (i).)
250 251 252 253	No information was uncovered to suggest that aquatic plant extracts could cause toxicity to plants, soil organisms, or higher animals (Original TAP Database Form, 1995).
253 254 255 256 257	The potential for over-harvesting of kelp/seaweed fields for production of aquatic plant extracts may be an environmental concern (Chalker-Scott, date unknown; Original TAP Database Form, 1995). However, no evidence has been found indicating that this potential impact is actually occurring.
257 258 259 260 261 262 263	Additionally, nutrients (i.e., nitrogen, phosphorous, and potassium) found in runoff from excessively or improperly applied fertilizers can cause excess algae growth in surface water (i.e., eutrophication). Excess algae can, in turn, use up oxygen in the water, potentially harming fish and other aquatic animals. No information has been found indicating that aquatic plant extracts are responsible for excessive nutrient runoff.

4 5	<b>Evaluation Question #6:</b> Is there potential for the petitioned substance to cause detrimental chemical interaction with other substances used in organic crop or livestock production? (From 7 U.S.C. § 6518
6	(m) (1).)
7 8 9 0	Based on the intended use of the substance, no information was uncovered to suggest that aquatic plant extracts could cause detrimental chemical interaction with other substances used in organic crop production.
1 2	Evaluation Question #7: Are there adverse biological or chemical interactions in the
2 3 4	agro-ecosystem by using the petitioned substance? (From 7 U.S.C. § 6518 (m) (5).)
5	Some aquatic plant extracts produced by sodium hydroxide extraction may contain excess sodium. If this
6 7 8	product is applied at high concentrations, plants may suffer (e.g., foliar damage, decreased plant growth) due to their sensitivity to salt (Original TAP Database Form, 1995).
9 0 1 2	Aquatic plant extract products that are not neutralized may have a high pH, which has a potential to adversely affect plants. However, no evidence of aquatic plant extract products adversely affecting crops has been identified. No studies have been found comparing the effectiveness of high-pH versus neutralized alkali extracts (Henry, 2005).
3 4 5 6	<b>Evaluation Question #8:</b> Are there detrimental physiological effects on soil organisms, crops, or livestock by using the petitioned substance? (From 7 U.S.C. § 6518 (m) (5).)
7	As previously mentioned, aquatic plant extracts can be used as a source of micronutrients and growth
8	promoters and potash <sup>2</sup> ; as a rooting solution for transplants and cuttings; for cold hardiness in tomato
)	plant, citrus fruits, and cabbage; and to reduce pest damage (including nematodes in tomato and okra;
) l	mites in strawberries, peaches and apples) (Original TAP Database Form, 1995).
2	Aquatic plant extracts are unlikely to create unacceptable changes in soil temperature, water availability,
	pH levels, nutrient availability, or salt concentration. However, if aquatic plant extracts produced by
	sodium hydroxide extraction (which may contain excess sodium) are applied at high concentrations, the
	plants may suffer from foliar damage (burn) due to their sensitivity to salt (Original TAP Database Form, 1995). Additionally, as previously stated, aquatic plant extract products that are not neutralized may have
	a high pH, and this has the potential to affect plants adversely. For example, micronutrients (i.e., copper, iron, manganese, zinc, and chloride) tend to be less available in soils with high pH.
	Evaluation Question #9: Is there a toxic or other adverse action of the petitioned substance or its
	breakdown products? (From 7 U.S.C. § 6518 (m) (2).)
	Based on their intended use, aquatic plant extracts and their breakdown products are unlikely to have toxic
	or other adverse actions. Aquatic plant extracts are composed primarily of natural plant constituents, such
	as proteins, lipids, glucides, amino acids, and vitamins (Original TAP Database Form, 1995). These
	components are generally biodegradable and nontoxic, and many can be utilized by organisms.
	<b>Evaluation Question #10:</b> Is there undesirable persistence or concentration of the petitioned substance
	or its breakdown products in the environment? (From 7 U.S.C. § 6518 (m) (2).)
	According to Tooby (2003), aquatic plant extracts appear to be biodegradable. Thus, the products and their breakdown products are unlikely to have an undesirable persistence in the environment.

<sup>&</sup>lt;sup>2</sup> Potash refers to a variety of mined and manufactured salts, all containing the element potassium in water-soluble form (USGS, 2001). January 31, 2006

314 315 316	Evaluation Question #11: Is there any harmful effect on human health by using the petitioned substance? (From 7 U.S.C. § 6517 (c) (1) (A) (i), 7 U.S.C. § 6517 (c) (2) (A) (i)) and 7 U.S.C. § 6518 (m) (4).)
317 318 319 320 321 322 323	Other than the potential for lung irritation due to inhalation of the soluble powder form of aquatic plant extract, no adverse effects on human health from the intended use were identified. According to the Material Safety Data Sheet (MSDS) for Necessary Organics' SeaMix Liquid Fertilizer, presented as an attachment to the Original 1995 TAP Database Form, the liquid form poses "no known health hazards" (Original TAP Database Form, 1995). According to Tooby (2003), plant extracts are believed to be associated with likely low adverse effects on human health.
324 325 326 327 328 329 330 331	However, the potential for sensitivity or intolerance to seaweed and, therefore, aquatic plant extracts, may exist. For example, it may be possible for some people to be sensitive to monosodium glutamate (MSG) <sup>3</sup> , which is found naturally in protein-rich food products, including seaweed. There are some reports that mild, temporary reactions to MSG may occur in a small portion of the population. Nonetheless, MSG is not considered an allergen by the American College of Allergy, Asthma and Immunology (IFIC, 1997). Additionally, the U.S. Food and Drug Administration has found no evidence to suggest any long-term, serious health consequences from consuming MSG (IFIC, 1997).
332 333 334	<u>Evaluation Question #12:</u> Is there a wholly natural product which could be substituted for the petitioned substance? (From 7 U.S.C. § 6517 (c) (1) (A) (ii).)
334 335 336 337 338 339 340	Natural products that may substitute for aquatic plant extract in its intended use include nonsynthetic fertilizers and soil amendments, including manure, blood meal, bone meal, compost, feather meal, guano, and other nonsynthetic animal or plant products. Also, aquatic plant extracts produced by nonsynthetic means (as described in Evaluation Question #1) are potential substitutes for synthetically produced products.
341 342 343 344	Although manure is a complete fertilizer, it lacks vital nutrients. An NPK of 1-1-1 is typical for manure <sup>4</sup> , though the nutrient content of manure is dependent upon the diet and species of the animal that produced it. Fresh manure contains the highest level of nutrients needed for plant growth.
345 346 347	<u>Evaluation Question #13:</u> Are there other already allowed substances that could be substituted for the petitioned substance? (From 7 U.S.C. § 6518 (m) (6).)
348 349 350	According to the NOP rule (7 CFR 205.601(j)), other allowed plant or soil amendments that could be substitutes for aquatic plant extracts, based on the intended uses, include the following:
351 352	• <b>Humic acids</b> - naturally occurring deposits, water and alkali extracts only;
353 354 355	• <b>Micronutrients</b> - not to be used as a defoliant, herbicide, or desiccant. Those made from nitrates or chlorides are not allowed. Soil deficiency must be documented by testing; and
356 357 358	• Liquid fish products - can be pH adjusted with sulfuric, citric or phosphoric acid. The amount of acid used shall not exceed the minimum needed to lower the pH to 3.5; (NOP, 2005).
359 360 361	<b>Evaluation Question #14:</b> Are there alternative practices that would make the use of the petitioned substance unnecessary? (From 7 U.S.C. § 6518 (m) (6).)
362 363 364 365	Potential alterative practices include recycling compost or other sources of nutrients. Alternative soil building practices include using cover crops, compost and/or manures, reducing tillage, avoiding traffic on wet soils, and maintaining soil cover with plants and/or mulches (Feenstra et al., 1997).

<sup>4</sup>See: http://www.ext.vt.edu/departments/envirohort/factsheets2/fertilizer/jan89pr6.html

<sup>&</sup>lt;sup>3</sup> MSG is a sodium salt of the amino acid glutamate (IFIC, 1997).

366	REFERENCE LIST
367 368 369	Abetz, P. 1980. Seaweed extracts: Have they a place in australian agriculture or horticulture? <i>The Journal of the Australian Institute of Agricultural Science</i> 46:23-29 (as cited in Tourte, 1997).
<ul><li>370</li><li>371</li><li>372</li><li>272</li></ul>	Acadian Seaplants Limited. Undated. ASL <sup>™</sup> kelp meal for soil application. Available Online at: <u>http://www.acadianseaplants.com/soilapplication.html</u> . Last Accessed: January 24, 2006.
<ul> <li>373</li> <li>374</li> <li>375</li> <li>376</li> <li>377</li> <li>378</li> </ul>	Allen, V.G., K.R. Pond, K.E. Sakers, J.P. Fontenot, C.P. Bagley, R.L. Ivy, R.R Evans, R.E. Schmidt, J.H. Fike, X. Zhang, J.Y. Ayad, C.P. Brown, M.F. Millert, J.L. Montgomery, J. Mahan, D.B. Wester, C. Melton. 2001. Tasco: Influence of a brown seaweed on antioxidants in forages and livestock – A review. <i>In J. Anim. Sci.</i> 79(E. Supple.): E21 – E31.
<ul> <li>379</li> <li>380</li> <li>381</li> <li>382</li> </ul>	Baldwin, K. R. 2001. Soil fertility for organic farming. Available Online at: http://www.ncsu.edu/organic_farming_systems/news/soil_fertility.PDF. Last Accessed: January 23, 2006.
383 384 385	Baker, B. 1996. Plant nutrition from the sea: Marine products can be used to supplement soil nutrients. <i>Farmer to Farmer</i> 16: page numbers not listed. Available Online at: <a href="http://www.noamkelp.com/nutrition.html">http://www.noamkelp.com/nutrition.html</a> . Last Accessed: January 23, 2006.
386 387 388 389	Canadian General Standards Board (CGSB). 2004. Organic Production Systems, Part 3 – Permitted Substances lists DRAFT. Available Online at: <a href="http://www.cog.ca/documents/OrganicStandard2004.3.pdf">http://www.cog.ca/documents/OrganicStandard2004.3.pdf</a> . Last Accessed: January 24, 2006.
390 391 392 393	Chalker-Scott, L. Date unknown. The myth of curative kelp: "Seaweed extracts reduce disease, improve production, and increase stress resistance in landscape plants". Available Online at: http://www.puyallup.wsu.edu/~Linda%20Chalker-
394	Scott/Horticultural%20Myths_files/Myths/Seaweed%20extracts.pdf. Last Accessed: January 31, 2006.
395 396 397 398	Conrad, Brad. July 16, 2002. Letter to Bob Pooler of the National Organics Standards Board from Arcadian Sea Plants
399 400 401	Feenstra, G., C. Ingels, D. Campbell, D. Chaney, M.R. George, and E. Bradford. 1997. What is sustainable agriculture? Available Online at: <u>http://www.sarep.ucdavis.edu/concept.htm</u> . Last Accessed: January 24, 2006.
402 403 404 405	Hall, B. 1997. Nonconventional Soil Amendments. <i>Appropriate Technology Transfer for Rural Areas</i> . Fayetteville, AR (as cited in Baldwin, 2001).
406 407 408 409	Hall, B. and P. Sullivan. 2001. Alternative Soil Amendments. <i>Appropriate Technology Transfer for Rural Areas</i> . Available Online at: <u>http://attra.ncat.org/attra-pub/PDF/altsoil.pdf</u> . Last Accessed: January 24, 2006.
410 411 412 413	Henry, Eric C. 2005. Report of Alkaline Extraction of Aquatic Plants. Available Online at: <u>http://www.omri.org/AdvisoryCouncil/Aquatic_plant_extracts-2004-02-14.pdf</u> . Last Accessed January 24, 2006.
414 415 416 417	International Food Information Council Foundation (IFIC). 1997. Everything you need to know about glutamate and monosodium glutamate. Available Online at: <u>http://ific.nisgroup.com/publications/brochures/upload/Everything-You-Need-To-Know-About-</u> Clutamate And Manasadium Clutamate add. Last Accessed: January 30, 2006
417 418 419 420	<u>Glutamate-And-Monosodium-Glutamate.pdf</u> . Last Accessed: January 30, 2006. Kuisma, P. 1989. The effect of foliar application of seaweed extract on potato. <i>Journal of Agricultural Science in Finland</i> 61: 371-377 (as cited in Tourte 1997).

421	
422	Laboratoires GOEMAR, S.A. undated. Technical document on Product GA 14. Provided as an attachment
423	to the Original TAP Database Form, 1995. (Fax dated from Agrimar Corp on January 12, 1995).
424	
425	McGeary, D.J. and W.E. Birkenhead. 1984. Effect of Seaweed Extract on Growth and Yield of Onions. <i>The</i>
426 427	Journal of the Australian Institute of Agricultural Science 50: 49-50 (as cited in Tourte 1997).
428	North Central Regional's NCR-103 Committee on Nontraditional Soil Amendments and Growth
429	Stimulants (NCR). 2004. Non-conventional soil additives: products, companies, ingredients, and claims.
430	Available Online at: http://www.soils.wisc.edu/extension/hottopics/nonconventional.pdf. Last
431	Accessed: January 31, 2006.
432	Necessary Ownersing 1001 Material Safety Data Chart on San Min Liquid Fartilizary Provided on an
433 434 435	Necessary Organics. 1991. Material Safety Data Sheet on SeaMix Liquid Fertilizer. Provided as an attachment to the Original TAP Database Form, 1995.
436	Nelson, W.R. and J. Van Staden. 1984. The effect of seaweed concentrate on growth of nutrient-stressed
437	greenhouse cucumbers. <i>HortScience</i> 19(1): 81-82 (as cited in Tourte, 1997).
438	
439	Nishimura, N., J. Zhang, M. Abo, A. Okubo, and S. Yamazaki. 2001. Simultaneous determination of betaines and
440	free amino acids in higher plants by capillary electrophoresis. Analytical Sciences 17(supplement): 929-932.
441	Our anis The day Association 2002. Commention Analysis of the United Cluster Nethersh One aris December (7
442	Organic Trade Association. 2002. Comparative Analysis of the United States National Organic Program (7
443	CFR 205) and the European Union Organic Legislation (EEC2092/91 & Amendments). Available Online at:
444 445	http://www.ota.com/pics/documents/NOPEUunifiedreport.pdf. Last Accessed: January 30, 2006.
446	OMRI 2005. OMRI Brand Name Product List. Organic Materials Review Institute. Eugene, Oregon.
447	January 2005.
448	Juliuly 2000.
449	Original TAP Review. 1999. Phosphoric Acid (Livestock). From NOSB Materials Database.
450	
451	Original TAP Database Form. 1995. Aquatic Plant Extracts. From NOSB Materials Database.
452	
453	Sideman, E. October 12, 2004. Letter to NOSB and NOP regarding comment on aquatic plant extracts.
454	
455	Smittle, D. 1991. Seaweed comes ashore. <i>Fine Gardening Magazine</i> . Available Online at:
456	http://www.casoils.com/seaweed_comes_ashore.html. Last Accessed: January 23, 2006.
457	
458	3R Lighting. 2005. Seaweed fertilizer. Available Online at:
459	http://www.3rlighting.com/organic/fertilizer/seaweed.html. Last Accessed: January 24, 2006.
460 461	Tooby, T. E. 2003. Registration opportunities for natural products versus synthetic plant stress signals (or
462	plant activators) for crop protection. Available Online at: <u>http://www.ibma.ch/pdf/200311_paper.pdf</u> .
463	Last Accessed: January 24, 2006.
464	Last Accessed. January 24, 2000.
465	Tourte, Laura. 1997. Kelp extract and fish powder sprays on organically grown processing tomatoes.
466	Organic Farming Research Foundation's Information Bulletin 4: 6-7 and 9.
467	
468	U.S. Department of Agriculture's National Organic Program (NOP). 2005. National List: Regulatory Text.
469	Subpart G – Admistrative. The National List of Allowed and Prohibited Substances. § 205.600 Evaluation
470	criteria for allowed and prohibited substances, methods, and ingredients. Available Online at:
471	http://www.ams.usda.gov/nop/NOP/standards/ListReg.html. Last Accessed: January 24, 2006.
472	
473	U.S. Environmental Protection Agency (EPA). 1998. Certain plant regulators, cytokinins, auxins,
474	gibberellins, ethylene, and pelargonic acid; tolerance exemptions. <i>Federal Register</i> 63(205): 56882-56886.

- Available Online at: http://www.epa.gov/fedrgstr/EPA-PEST/1998/October/Day-23/p28360.htm. Last
  Accessed: January 31, 2006.
- 476 Accessed: January 31, 200477
- 478 U.S. Geological Survey (USGS). 2001. Potash statistics and information. Available Online at:
- 479 <u>http://minerals.usgs.gov/minerals/pubs/commodity/potash/</u>. Last Accessed: January 30, 2006.
- 480
- 481 Verkleij, F.N. 1992. Seaweed extracts in agriculture and horticulture: a review. *Biological Agriculture and*
- 482 *Horticulture* 8: 309-324 (as cited in Tourte, 1997).

483 484